

MATH327: StatMech and Thermo

Wednesday, 12 March 2025

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Something to consider

You may have heard that the first and second laws of thermodynamics rule out the existence of perpetual-motion machines.

How can we see this at play in thermodynamic cycles?

Recap

Carnot cycle

$$\text{Efficiency } \eta = \frac{W_{\text{out}} - W_{\text{in}}}{Q_{\text{in}}}$$

$$\text{First law: } 0 < \eta \leq 1$$

"can't win"

Today

Therm. cycle wrap-up

Grand - canonical ensemble

Carnot cycle efficiency

$$\text{Last time: } W_{AB} = P_A V_A \log\left(\frac{V_A}{V_B}\right) < 0 \rightarrow W_{\text{out}}$$

$$Q_{AB} = -W_{AB} > 0 \rightarrow Q_{\text{in}}$$

$$Q_{BC} = 0$$

$$W_{BC} = \frac{3}{2} P_A V_A \left[\left(\frac{V_B}{V_A}\right)^{2/3} - 1 \right] < 0 \rightarrow W_{\text{out}}$$

3) C → D

$$W_{CD} = - \int_{V_C}^{V_D} \frac{N T_L}{V} dV = P_A V_A \left(\frac{T_L}{T_H} \right) \log \left(\frac{V_C}{V_D} \right)$$

$$= P_A V_A \left(\frac{V_B}{V_C} \right)^{2/3} \log \left(\frac{V_B}{V_A} \right) > 0 \rightarrow W_{in}$$

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$$Q_{CD} = -W_{CD} < 0 \rightarrow Q_{out}$$

4) D → A $Q_{DA} = 0$

$$W_{DA} = \frac{3}{2} N (T_H - T_L) = -W_{BC} > 0 \rightarrow W_{in}$$

$$\eta = \frac{W_{out} - W_{in}}{Q_{in}} = \frac{-W_{AB} - W_{BC} - W_{CD} - W_{DA}}{-W_{AB}} = 1 + \frac{W_{CD}}{W_{AB}}$$

$$W_{out} = -W_{AB} - W_{BC} > 0$$

$$W_{in} = W_{CD} + W_{DA} > 0$$

$$Q_{in} = Q_{AB} = -W_{AB} > 0$$

$$\eta = 1 + \frac{P_A V_A \left(\frac{V_B}{V_C} \right)^{2/3} \log \left(\frac{V_B}{V_A} \right)}{P_A V_A \log \left(\frac{V_A}{V_B} \right)} = 1 - \frac{T_L}{T_H} \quad T_L < T_H$$

So $0 < \eta < 1$

$$\frac{T_L}{T_H} \rightarrow 1$$

$$\frac{T_L}{T_H} \rightarrow 0$$

abs. zero or infinitely hot

General results

1) $\frac{T_L}{T_H} \rightarrow 1$ always means $\eta \rightarrow 0$
No heat flow to do work

2) $\frac{T_H}{T_L}$ increasing \rightarrow more efficient

3) Recall $W_{out} - W_{in} = Q_{in} - Q_{out}$ From first law

$$\eta = \frac{Q_{in} - Q_{out}}{Q_{in}} = 1 - \frac{Q_{out}}{Q_{in}} \quad \begin{array}{l} \text{to cold res} \\ \text{From hot res.} \end{array} \quad Q = T dS$$

$$= 1 - \frac{T_L \Delta S_{out}}{T_H \Delta S_{in}}$$

After each iteration, same S_A in system
 ΔS_{in} removed from hot res.
 ΔS_{out} added to cold res.

Second law: $\Delta S = \Delta S_{out} - \Delta S_{in} \geq 0$

$$\frac{\Delta S_{out}}{\Delta S_{in}} = \frac{Q_{out}/T_L}{Q_{in}/T_H} \geq 1$$

$$\frac{Q_{out}}{Q_{in}} \geq \frac{T_L}{T_H} > 0$$

\therefore For any therm cycle

$$\eta = 1 - \frac{Q_{out}}{Q_{in}} \leq 1 - \frac{T_L}{T_H} < 1$$

"can't break even"

Carnot cycle gives max. possible η

Not used - too slow!

Reverse cycle - put in work to remove heat
From cold res.
 \rightarrow refrigerator

Coefficient of performance (COP)

$$\frac{Q_{in}}{W_{in} - W_{out}} = \frac{Q_{in}}{Q_{out} - Q_{in}} = \frac{1}{\frac{Q_{out}}{Q_{in}} - 1}$$

Now $\left. \begin{array}{l} Q_{in} = T_L \Delta S_{in} \\ Q_{out} = T_H \Delta S_{out} \end{array} \right\} \frac{Q_{out}}{Q_{in}} \geq \frac{T_H}{T_L}$ by second law

$$COP \leq \frac{1}{\frac{T_H}{T_L} - 1} = \frac{T_L}{T_H - T_L}$$

Typical COP ~ 5 , maximized by Carnot cycle

Allow both heat and particle exchange with reservoir
 \swarrow fix T \nwarrow what else?

Recall $\frac{1}{T} = \left. \frac{\partial S}{\partial E} \right|_N$

$-T \left. \frac{\partial S}{\partial N} \right|_E = \mu$ is chemical potential

The grand-canonical ensemble

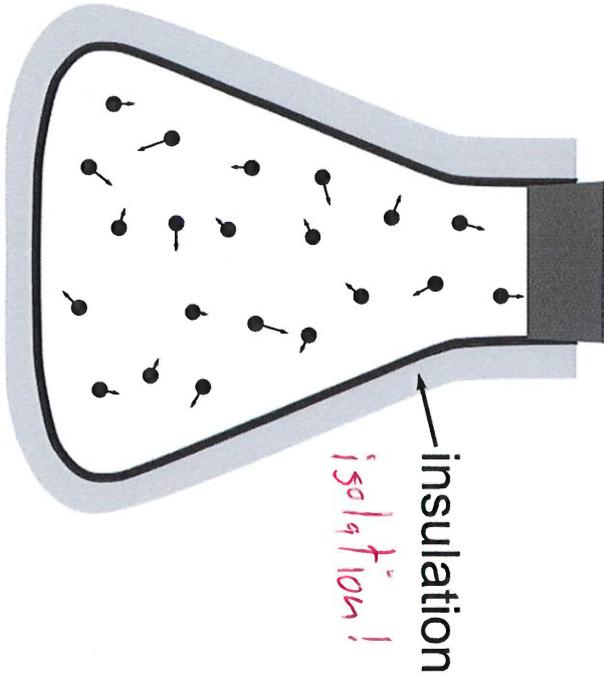
is characterized by fixed T and μ
through energy and particle exchange with particle res.

$\mu = -T \left. \frac{\partial S}{\partial N} \right|_E$ is intensive, dimensions of E & T

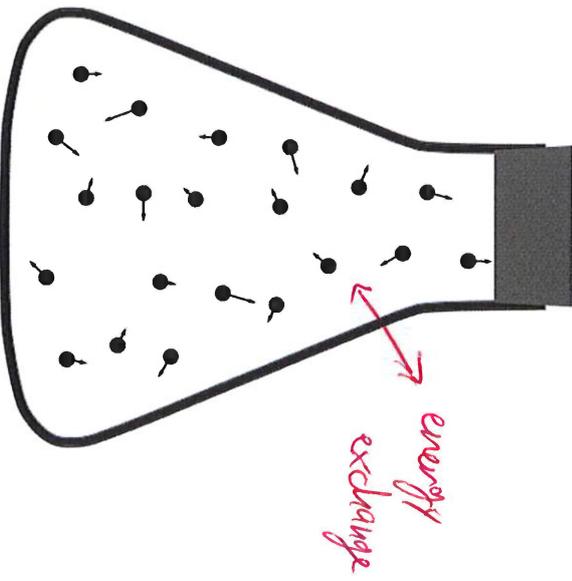
For $T > 0$ "natural" systems

more particles \rightarrow more entropy even with fixed E

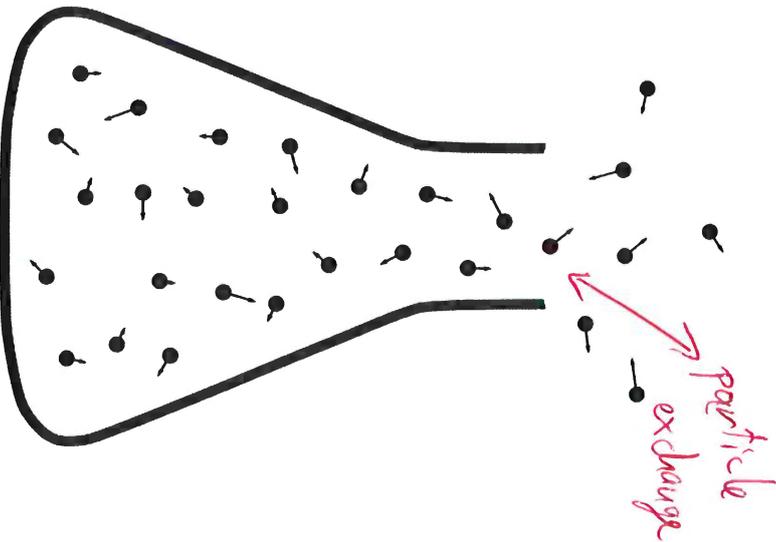
$$\left. \frac{\partial S}{\partial N} \right|_E > 0 \rightarrow \mu < 0$$



Microcanonical
(const. N E)



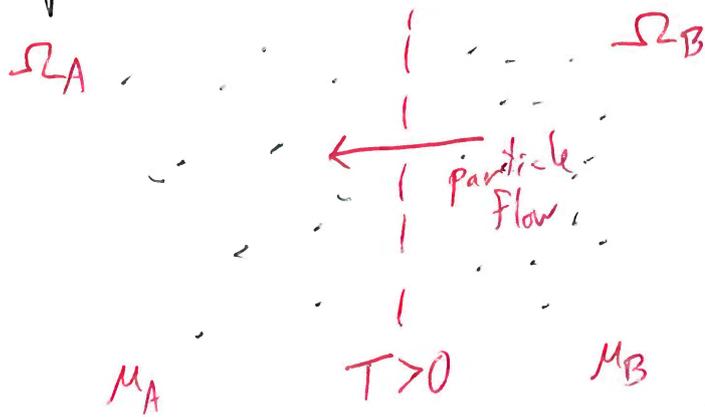
Canonical
(const. N T)



Grand Canonical
(const. μ T)

Sign in choice to aid intuition

Consider particle flow with $\mu_A < \mu_B < 0$



same $\rightarrow \frac{\partial S_A}{\partial N_A} > \frac{\partial S_B}{\partial N_B} > 0$

$$\Delta N_A = -\Delta N_B$$

$$\Delta S = \Delta S_A + \Delta S_B = \frac{\partial S_A}{\partial N_A} \Delta N_A + \frac{\partial S_B}{\partial N_B} \Delta N_B \geq 0 \text{ by second law}$$

$$\Delta N_A \left[\frac{\partial S_A}{\partial N_A} - \frac{\partial S_B}{\partial N_B} \right] \geq 0 \rightarrow \Delta N_A > 0$$

Particles flow from larger μ (Ω_B)
to smaller μ (Ω_A)
like heat flow and T

N_i and E_i depend on micro-state w_i

Want p_i free from dependence on reservoir

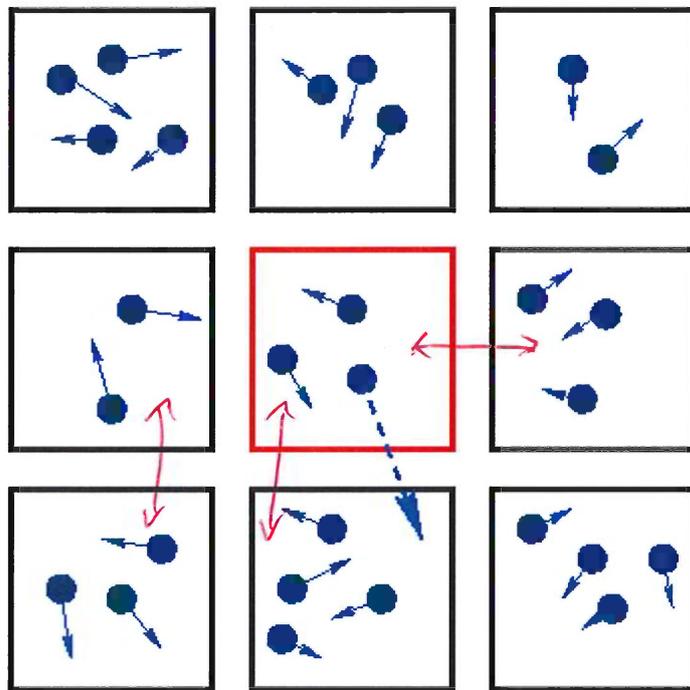
Adapt replica trick

$$E_{\text{tot}} = E + E_{\text{res}} = \sum_{r=1}^R E_r$$

$$N_{\text{tot}} = N + N_{\text{res}} = \sum_r N_r$$

Ω has M micro-states $w_i = w_1, w_2, \dots, w_M$
with energy E_i and N_i indist'able particles

$$\Omega_{\text{tot}} = \Omega \otimes \Omega_{\text{res}} \rightarrow \mathcal{R} \text{ spires of } \Omega$$



Recall occupation numbers n_i
↳ replicas in W_i

and probabilities $P_i = n_i/R$

$$\sum_{i=1}^M n_i = R$$

$$\sum_i P_i = 1$$

$$E_{\text{tot}} = \sum_i n_i E_i$$

$$N_{\text{tot}} = \sum_i n_i N_i = R \sum_i P_i N_i$$

Compute (intensive) T and μ of micro-canonical Ω_{tot}
Need therm. equil. \rightarrow maximizing S_{tot} with constraints

Same $M_{\text{tot}} = \binom{R}{n_1} \binom{R-n_1}{n_2} \dots = \frac{R!}{n_1! n_2! \dots n_M!}$

$$S_{\text{tot}} = -R \sum_i P_i \log P_i \quad n_i \gg 1$$

Another Lagrange multiplier

$$\bar{S} = -R \sum_i P_i \log P_i + \alpha \left(\sum_i P_i - 1 \right) - \beta (R \sum_i P_i E_i - E_{\text{tot}}) + \gamma (R \sum_i P_i N_i - N_{\text{tot}})$$

$$\frac{\partial \bar{S}}{\partial P_k} = 0 = -R (\log P_k + 1) + \alpha - \beta R E_k + \gamma R N_k$$

$$\log P_k = -1 + \frac{\alpha}{R} - \beta E_k + \gamma N_k$$

$$P_k = \frac{\exp(-\beta E_k + \gamma N_k)}{\exp(1 - \frac{\alpha}{R})} = \frac{1}{Z_g} e^{-\beta E_k + \gamma N_k}$$

Impose constraints

$$1 = \sum_i p_i = \frac{1}{Z_g} \sum_i e^{-\beta E_i + \gamma N_i}$$

$$Z_g = \sum_i e^{-\beta E_i + \gamma N_i} = Z_g(\beta, \gamma)$$

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grand-canonical partition function

T and μ come from entropy

$$S_{\text{tot}} = -R \sum_i p_i \log \left(\frac{1}{Z_g} e^{-\beta E_i + \gamma N_i} \right)$$

$$= -R \sum_i p_i (-\log Z_g - \beta E_i + \gamma N_i)$$

$$= R \log Z_g + \beta E_{\text{tot}} - \gamma N_{\text{tot}}$$

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$$\frac{1}{T} = \frac{\partial S}{\partial E} \Big|_W = R \left(\frac{\partial \beta}{\partial E} \frac{\partial}{\partial \beta} \log Z_g + \frac{\partial \gamma}{\partial E} \frac{\partial}{\partial \gamma} \log Z_g \right) + \beta + E \frac{\partial \beta}{\partial E} - N \frac{\partial \gamma}{\partial E}$$

$$1) \frac{1}{Z_g} \sum_i \frac{\partial}{\partial \beta} e^{-\beta E_i + \gamma N_i} = \frac{1}{Z_g} \sum_i (-E_i) e^{-\beta E_i + \gamma N_i} = -\sum_i p_i E_i = \frac{-E}{R}$$

$$2) \frac{1}{Z_g} \sum_i \frac{\partial}{\partial \gamma} e^{-\beta E_i + \gamma N_i} = \sum_i p_i N_i = \frac{N}{R}$$

$$\frac{1}{T} = -E \frac{\partial \beta}{\partial E} + N \frac{\partial \gamma}{\partial E} + \beta + E \frac{\partial \beta}{\partial E} - N \frac{\partial \gamma}{\partial E} = \beta \quad \checkmark$$

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Similarly

$$-\frac{\mu}{T} = \frac{\partial S}{\partial N} \Big|_E = R \left(\frac{\partial \beta}{\partial N} \left(\frac{-E}{R} \right) + \frac{\partial \gamma}{\partial N} \left(\frac{N}{R} \right) \right) + E \frac{\partial \beta}{\partial N} - \gamma - N \frac{\partial \gamma}{\partial N}$$

$$\text{so } \gamma = \frac{\mu}{T}$$

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Putting things together, we have derived
the micro-state probabilities

$$P_i = \frac{1}{Z_g} e^{-E_i/T + \mu N_i/T} = \frac{1}{Z_g} e^{-\beta(E_i - \mu N_i)}$$

$$Z_g(T, \mu) = \sum_i e^{-(E_i - \mu N_i)/T}$$

Both E_i and N_i fluctuate

Reservoir unknowable apart from fixing T and μ

~~Predict~~

Preview: From Z_g predict

entropy $S(T, \mu)$

internal energy $\langle E \rangle(T, \mu)$

particle number $\langle N \rangle(T, \mu)$